M.Sc. 4th Semester Examination, 2021

CHEMISTRY

(Organic Chemistry Special)

Paper: CHEM 403E

Course Id: 41453

Time: 2 hours Full Marks: 40

The figures in the margin indicate full marks

Candidates are required to give their answers in their own words as far as practicable

1. Answer *any five* of the following questions:

 $2 \times 5 = 10$

- (a) What is iodine number?
- (b) There are two ways to count the position of double bond in a fatty acid, delta, and omega. Using both methods, classify arachidonic acid.

- (c) State the basic differences between Vibrational Circular Dichroism (VCD) and Optical Rotation Dispersion (ORD).
- (d) Write the structure of the C2 epimer of glucose.
- (e) Name an instrument used in the determination of secondary structure of a protein.
- (f) What are the advantages of solid phase peptide synthesis by fluorescamine method?
- (g) Mention two advantages of automated oligo-nucleotide synthesis using phosphotriester approach.
- 2. Answer *any four* of the following questions:

 $5 \times 4 = 20$

- (a) What is essential fatty acid? Give an example of omega-6 fatty acid. Between *cis* and *trans* fats which one is good for our health and why? 2+1+2=5
- (b) The specific rotation of α -D-glucose is +112° and β -D-glucose is +18.7° where as the specific rotation of the constant equilibrium mixture is +52.7°. Calculate the percentage

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composition of α and β anomers at equilibrium. How might conformational factors account for this result?

- (c) (i) The bond between C1 and C2 in α -glucopyranose is oxidised by HIO₄ more rapidly than the β -isomer explain.
- (ii) Glucose, mannose and fructose give identical osazone explain.
- (d) (i) How will you distinguish 2-deoxyglucose from 3-deoxyglucose?
- (ii) A monosaccharide is treated with HCN, the product hydrolysed and reduced to a carboxylic acid by heating with HI and P. What carboxylic acid is formed if the monosaccharide is (I) D-glucose (II) D-mannose (III) D-fructose?
- (e) (i) Mention two demerits of phosphotriester method for the synthesis of gene.
- (ii) What is the purpose of hydrogenation process of naturally occurring edible oil? Is it good or bad for our health? 2+(2+1)=5
- (f) Furnish a schematic diagram for the solid phase peptide synthesis by fluorescamine method.
- 3. Answer *any one* of the following questions:

 $1 \times 10 = 10$

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- (a) (i) Fluid mosaic model allows lateral movement but not rotation through the membrane-explain.
- (ii) Which method is successfully applied in automated protein sequencing? Taking a suitable example, show the mechanism involved in this sequencing reaction. Mention the name of the intermediate and reagents involved. 3+(1+3+3)=10
- (b) (i) Write short notes on (A) vasopressin (B) haloketone rule and (C) sector rule.
- (ii) Show why aldaric acid derived from ribose and xylose would not be optically active.

(3+3+3)+1=10

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